

Total Synthesis of Geranylgeranylgeranyl Phosphate Enantiomers: Substrates for Characterization of 2,3-*O*-Digeranylgeranylgeranyl Phosphate Synthase

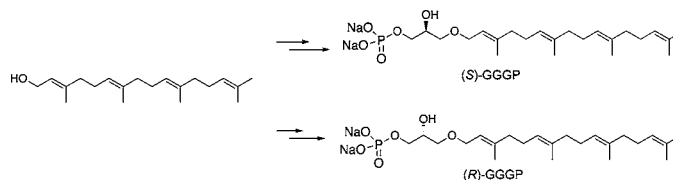
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ABSTRACT



To determine the enantioselectivity of (*S*)-2,3-di-*O*-geranylgeranylgeranyl phosphate synthase (DGGGPS) from the thermoacidophilic archaeon *Sulfolobus solfataricus*, we developed an efficient enantioselective route to the enantiomeric geranylgeranylgeranyl phosphates (*R*)-GGGP and (*S*)-GGGP. Previous routes to these substrates involved enzymatic conversions due to the lability of the polyprenyl chains toward common phosphorylation reaction conditions. The synthesis described herein employs a mild trimethyl phosphite/carbon tetrabromide oxidative phosphorylation to circumvent this problem. In contrast to previous results suggesting that only (*S*)-GGGP can act as the prenyl acceptor substrate, both (*R*)-GGGP and (*S*)-GGGP were found to be substrates for DGGGPS.

All living organisms have been classified into three primary kingdoms: archaeobacteria, eubacteria, and eukaryotes.¹ Although the membrane lipids in eubacteria and eukaryotes are composed of glyceryl esters of fatty acids, archaeobacterial membrane lipids contain isoprenyl glyceryl ethers.^{2,3} The core structures of archaeal membrane lipids, including diether and

bipolar tetraether lipids, contain fully reduced C₂₀ or C₂₅ prenyl groups.^{4,5} Biosynthesis of the isoprene moieties in the core lipids follows the mevalonate pathway used also by eubacteria and eukaryotes.^{6–8} The prenyl transfer reactions responsible for building the isoprenoid diphosphates, e.g., geranylgeranyl diphosphate (GGPP), are catalyzed by a

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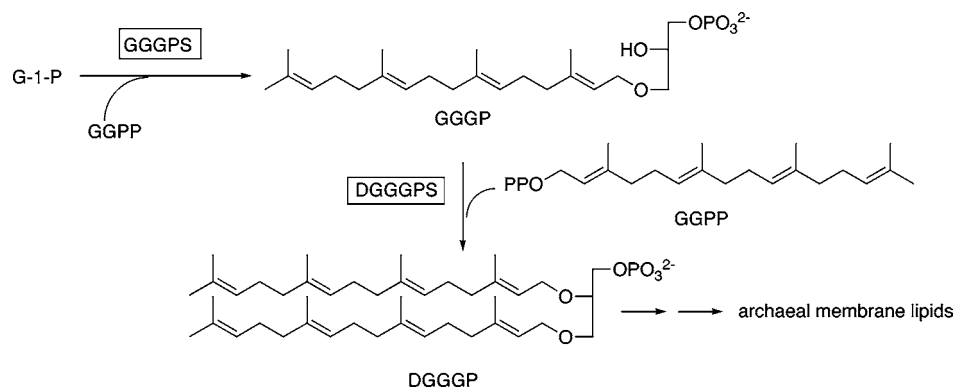


Figure 1. Biosynthesis of DGGGP, a key intermediate for archaeal membrane lipids. Key: G-1-P, (*S*)-glyceryl phosphate; GGGPS, (*S*)-GGGP synthase; DGGGPS, DGGGP synthase.

family of prenyltransferases.^{9–12} As shown in Figure 1, the biogenesis of the core structure of the archaeal membrane lipids starts with the prenyl transfer reaction catalyzed by (*S*)-3-*O*-geranylgeranylgeranyl glyceryl phosphate [(*S*)-GGGP] synthase, which selectively uses (*S*)-glyceryl phosphate as the prenyl acceptor (Figure 1). Then, the product is utilized as the presumed acceptor substrate for the biosynthesis of (*S*)-2,3-di-*O*-geranylgeranylgeranyl glyceryl phosphate (DGGGP), an advanced intermediate of archaeal membrane lipids.¹³

To date, only an enzyme-assisted synthesis of (*S*)-GGGP has been reported.¹¹ An enantiospecific chemical synthesis of both individual enantiomers of GGGP was required to validate this biosynthetic hypothesis, in part because of the acid-sensitive geranylgeranyl group of GGGP. To fully characterize the substrate selectivity of DGGGP synthase (DGGGPS), we developed a mild and effective route to the two GGGP enantiomers. The instability of GGGP indeed posed significant challenges, and the biological results with the enantiomers were unexpected.

The synthesis of (*S*)-GGGP (**8**) is summarized in Scheme 1. Treatment of the (*2E,6E,10E*)-geranylgeraniol **1** with Ph₃P/CBr₄ at 25 °C afforded geranylgeranyl bromide **2**.¹⁴ Next, (*S*)-solketal was alkylated with geranylgeranyl bromide by using KH as base, to give ether **3** in 73% yield.¹¹ The reported HCl/THF method to remove the acetamide¹³ resulted in a complex mixture containing the desired product in low yield. The desired diol **4** was thus prepared in 75% yield using *p*-TsOH in methanol.^{15,16} To obtain selective phos-

phorylation of the primary hydroxyl, we first tried a strategy involving protection of the secondary hydroxyl as a silyl ether. Thus, both hydroxyl groups of diol **4** were protected as TBS ethers (TBSCl, imidazole, anhydrous DMF).^{15,16} Subsequently, the more labile primary TBS ether was selectively removed at room temperature by using HF·Py (HF·Py/Py/THF = 1:2:5). Unfortunately, phosphorylation of the primary alcohol under standard conditions (dimethylphosphoryl chloride, *t*-BuOK, CH₂Cl₂)^{15,16} failed to give the desired product because of the lability of the polyene system.

A second strategy proved more successful. The use of the trimethyl phosphite/carbon tetrabromide oxidative phosphorylation method¹⁷ was deemed sufficiently mild to permit phosphorylation without damage to the geranylgeranyl moiety. Treatment of diol **4** with 1.1 equiv of CBr₄ and 1.2 equiv of P(OMe)₃ gave selective phosphorylation of the primary alcohol to give the protected phosphate **7**. Essentially no bisphosphate product was detected.

The next challenge in this synthesis was liberation of the free phosphate monoester from the protected triester. We first tried TMSBr, a standard deprotecting reagent for removal of methyl and ethyl groups in the synthesis of acyl-migration-prone lysophosphatidic acid derivatives.^{15,16} However, GGGP did not survive this strong Lewis acid. By using a solution of TMSBr in 2,4,6-trimethylpyridine (*sym*-collidine),^{18–20} we obtained the desired monophosphate in the acidic form. Titration with 1 N aq NaOH afforded (*S*)-GGGP (**8**) as the stabilized sodium salt.

To determine the enantioselectivity of DGGGPS, both the enantiomers (*S*)-GGGP (**8**) and (*R*)-GGGP (**12**) were re-

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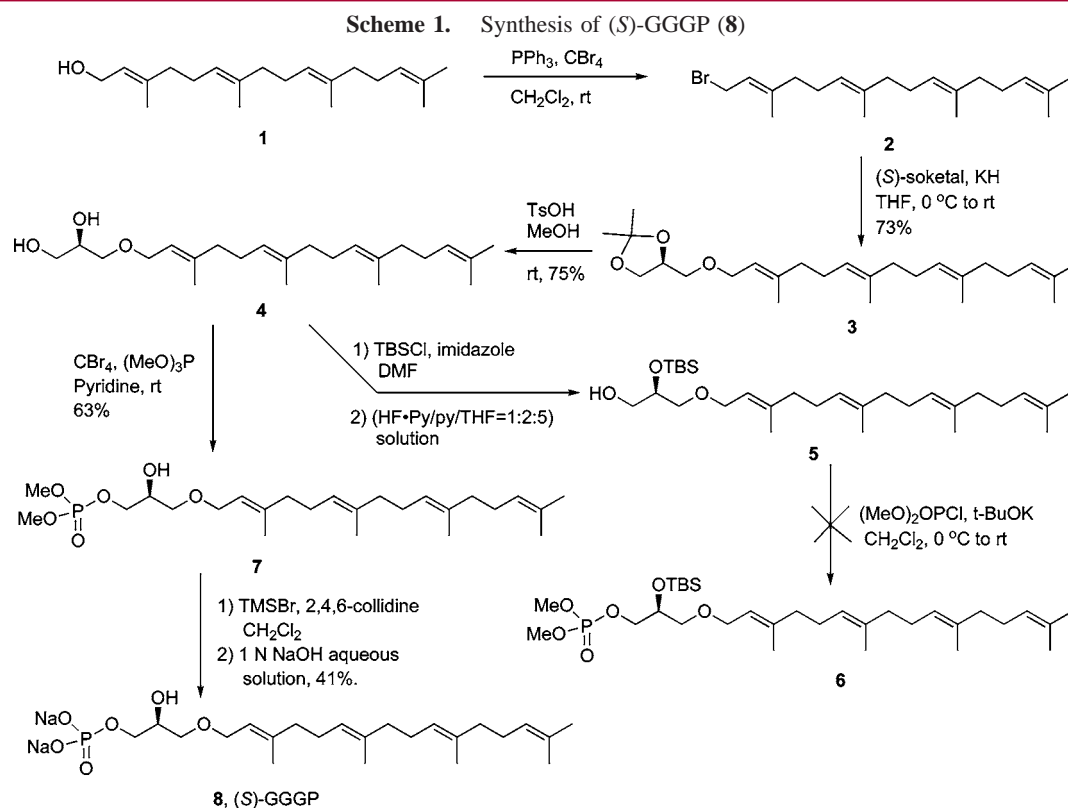
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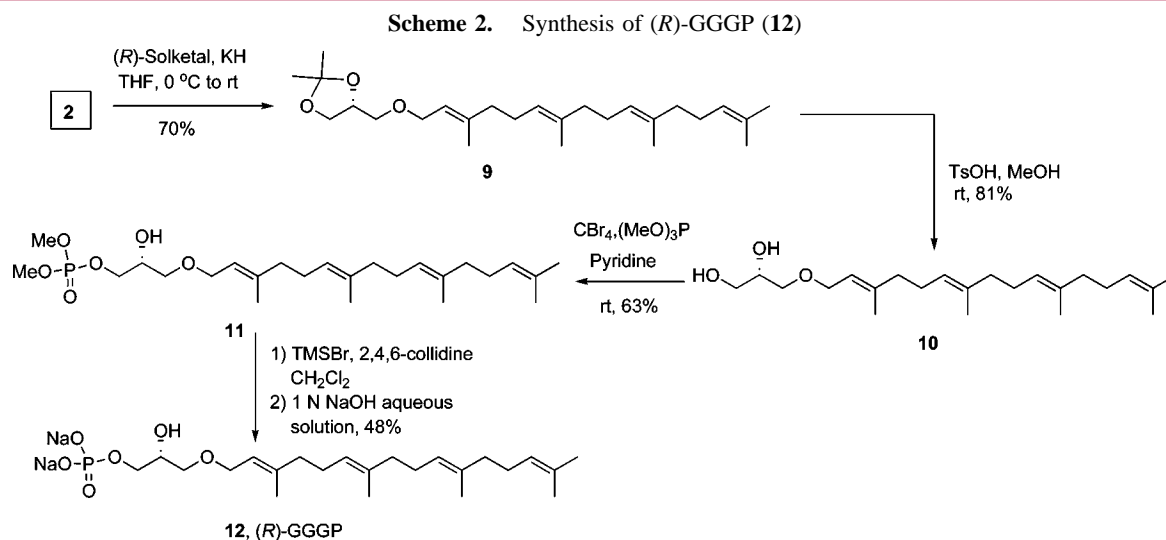
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quired. Starting with (*R*)-solketal, (*R*)-GGGP (**12**) was synthesized using the successful route as summarized in Scheme 2.

DGGGPS is a member of the UbiA prenyltransferase family that can catalyze the transfer of a prenyl group to its biological acceptor substrate (*S*)-GGGP (**8**). With the enantiomeric substrates (*R*)-GGGP (**12**) and (*S*)-GGGP (**8**) in hand, we determined the activity of DGGGPS toward each of these substrates. From the results of radio HPLC analysis (Figure 2) and reversed-phase TLC analysis (Figure 3), we

found that the DGGGP and presumably its enantiomer were formed in the reactions using (*S*)-GGGP (**8**) and (*R*)-GGGP (**12**), respectively. In these reactions, the starting reagent [^{14}C]-GGPP was formed first from [^{14}C]isopentenyl diphosphate and (*E,E*)-farnesyl diphosphate by the activity of GGPS. Then, [^{14}C]-GGPP was used as the prenyl donor substrate for DGGGPS. Thus, the results demonstrated that the C_{20} -prenyl group of GGPP could be transferred to either of the two GGGP enantiomers by the action of DGGGPS. (*S*)-GGGP seems to be marginally preferred in Figure 2,



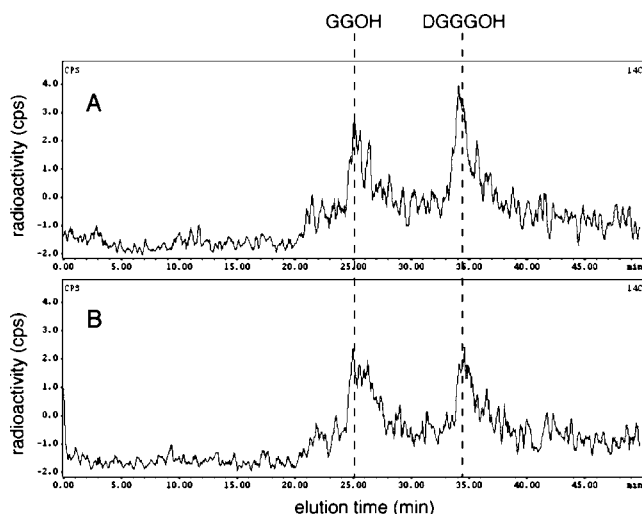


Figure 2. Radio HPLC analysis of 1-butanol extracts from the enzyme reaction with [^{14}C]-GGPP synthase and DGGGPs, using 0.5 nmol [^{14}C]-isopentenyl diphosphate, 0.5 nmol (*E,E*)-farnesyl diphosphate, 0.4 μmol (*S*)-GGGP (A), or (*R*)-GGGP (B) as substrates. Key: GGOH, geranylgeraniol; DGGGOH, digeranylgeranylgeranol.

whereas the (*R*)-enantiomer appeared to be preferred in Figure 3. These results are reproducible but qualitative; the GGGPs produced are quite labile. Nevertheless, to our surprise, both (*R*)-GGGP (**12**) and (*S*)-GGGP (**8**) were accepted at a comparable extent as substrates for DGGGPs. During the biosynthesis of archaeal membrane lipids, GGGPS catalyzes the transfer of prenyl groups from GGPP to (*S*)-glyceryl phosphate in the formation of (*S*)-GGGP (**8**), and the ether linkage between both (*S*)-GGGP (**8**) and another geranylgeranyl group is formed under the control of DGGGPs. GGGPSs are known to have strict substrate preferences: (*R*)-glyceryl phosphate is a very poor substrate.^{13,21} Thus, our results strongly suggest that the chirality of the archaeal membrane lipid is determined by GGGPS, not by DGGGPs. However, (*R*)-GGGP (**12**) and (*S*)-GGGP (**8**) will be important tools for more detailed analysis of the specific activity and enantioselectivity of DGGGPs in future studies.

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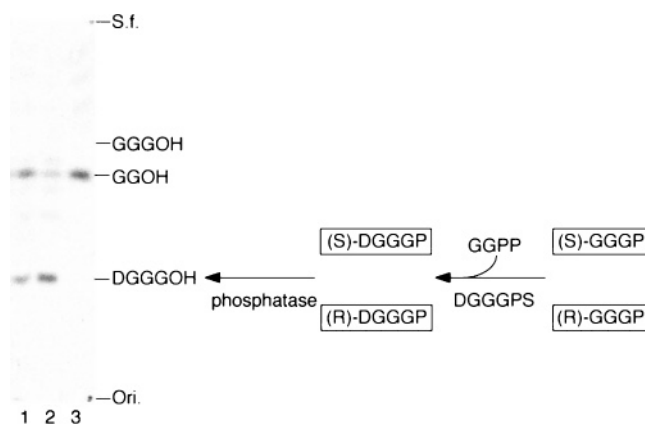


Figure 3. Autoradiogram of TLC from left to right: (1) (*S*)-GGGP; (2) (*R*)-GGGP; (3) without an acceptor substrate. Key: GGGOH, geranylgeranylgeranol; GGOH, geranylgeraniol; DGGGOH, digeranylgeranylgeranol; Ori, origin; S.f., solvent front.

In conclusion, (*S*)-GGGP and (*R*)-GGGP were each synthesized by a five-step procedure starting from the (*2E,6E,10E*)-geranylgeraniol and the appropriate enantiomer of solketal. A regioselective phosphorylation of diol **4** was achieved using $\text{CBr}_4/\text{P}(\text{OMe})_3$, and the instability problem of the geranylgeranyl group was circumvented by judicious selection of mild reaction conditions. The LKC18 reversed-phase TLC analysis and radio HPLC analysis have shown that the DGGGPs can catalyze the transfer of a prenyl group to the secondary hydroxy groups of both (*R*)-GGGP and (*S*)-GGGP.

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Supporting Information Available: Experimental procedures and characterization for all new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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